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The Spinels CuCr_2O_4 and CuRh_2O_4

W. A. DOLLASE^a AND H. ST. C. O'NEILL^b

^a*Department of Earth & Space Sciences, University of California, Los Angeles, CA 90095, USA, and* ^b*Research School of Earth Sciences, Australian National University, Canberra, ACT 0200, Australia. E-mail: dollase@ucla.edu*

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Abstract

Rietveld refinement of X-ray powder data measured for CuCr_2O_4 [copper(II) dichromium(III) tetraoxide] and CuRh_2O_4 [copper(II) dirhodium(III) tetraoxide] indicates that their structures at room temperature are centrosymmetric, $I4_1/amd$, rather than as previously reported, noncentrosymmetric $I\bar{4}2d$. Trial refinement in $I\bar{4}2d$ shows insignificant improvement in the agreement between observed and calculated intensities, and no statistically significant deviation from a centrosymmetric structure.

Comment

With sufficient concentration of non-spherical, so-called Jahn–Teller (JT) ions, such as Cu^{2+} or Mn^{3+} , the normally cubic $Fd\bar{3}m$ spinel structure may undergo a cooperative distortion to a tetragonal structure with either an elongated or compressed unit cell. A search of the Inorganic Crystal Structure Database (Bergerhoff, Hundt, Sievers & Brown, 1983) shows that all elongated JT spinels for which $c > a$ have the Mn_3O_4 hausmannite structure with space group $I4_1/amd$. However, all compressed JT spinels with $c < a$ are reported to have space group $I\bar{4}2d$, a noncentrosymmetric subgroup of $I4_1/amd$. Only two members of the $c < a$ group have been reported: CuCr_2O_4 and CuRh_2O_4 .

The room-temperature crystal structure of copper chromium spinel (Prince, 1957) was assigned the noncentrosymmetric space group $I\bar{4}2d$ over the centrosymmetric alternative, $I4_1/amd$, on the basis of a reported qualitatively better agreement between the ten observed and calculated neutron diffraction intensities. The identical $I\bar{4}2d$ structure was adopted for copper rhodium spinel (Dulac, 1969) without further discussion of the choice of space group.

A restudy of these two spinels using Rietveld refinement of X-ray powder diffraction data is reported here. Initial refinement in the previously suggested noncentrosymmetric space group yielded deviations from a centrosymmetric structure that were much smaller

than those reported by Prince (1957) and Dulac (1969), prompting trial refinement in the corresponding centrosymmetric space group. For CuCr_2O_4 , the pertinent R factors are: $R_p = 0.0279$, $R_{wp} = 0.0368$ and $R_B = 0.0387$ for a centrosymmetric model, and $R_p = 0.0279$, $R_{wp} = 0.0369$ and $R_B = 0.0385$ for the noncentrosymmetric case. Similarly, for CuRh_2O_4 these R factors are: $R_p = 0.0440$, $R_{wp} = 0.0603$ and $R_B = 0.0469$ for the centrosymmetric case, and $R_p = 0.0439$, $R_{wp} = 0.0603$ and $R_B = 0.0467$ for the noncentrosymmetric case. Due to the increased number of (atom position) variables, some small apparent improvement may be expected with a noncentrosymmetric model, but the differences here are seen to be minor.

The two additional positional parameters, $x(\text{Cr,Rh})$ and $x(\text{O})$, in $I\bar{4}2d$ refined to within 2.9 and 0.1 standard uncertainties of their symmetry-fixed $I4_1/amd$ values in the chromium spinel, and to within 2.2 and 0.6 standard uncertainties in the rhodium spinel. Furthermore, with a second, but less extensive, data set measured on the same rhodium spinel sample, the Rh and O positional variables refined to within 0.6 and 1.8 standard uncertainties of the centrosymmetric fixed values. Taken together, these data demonstrate that the apparent deviations from a centrosymmetric structure are not statistically significant and that the correct space group is, in both cases, $I4_1/amd$.

Trial refinement of the tetrahedral and octahedral site occupancies revealed no detectable deviation from full occupancy of Cu in the tetrahedral sites of both structures and full occupancy of Cr (or Rh) in the octahedral sites. Both may be therefore classified as normal spinels.

The refined atomic positional parameters for the centrosymmetric space group are given in Tables 1 and

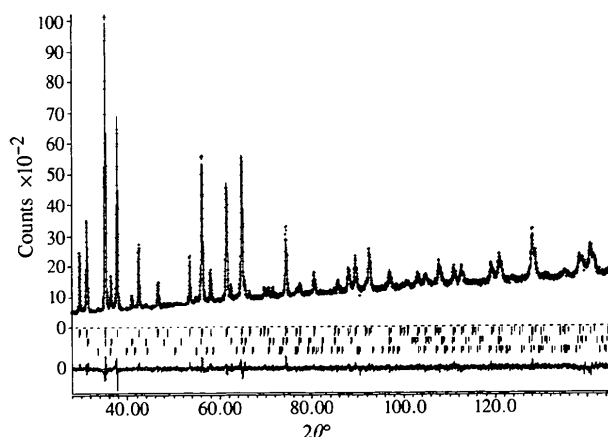


Fig. 1. Observed (crosses), calculated (solid line) and difference profiles from Rietveld analysis of the chromium spinel sample. Tick marks indicate the positions of the allowed reflections for CuCr_2O_4 (upper), CuCrO_2 (middle), and Cr_2O_3 (lower).

4. All standard uncertainties quoted are corrected for the effect of serial correlation of the data (Bérar & Lelann, 1991). The refined cell dimensions are in satisfactory agreement with the most recent literature values: $a = 6.0336(3)$, $c = 7.7824(6)$ Å for CuCr₂O₄ (Morris *et al.*, 1984) and $a = 6.175$, $c = 7.902$ Å for CuRh₂O₄ (Powder Diffraction File No. 41-402).

Interatomic distances and angles are listed for comparison. In common $Fd\bar{3}m$ spinels, the tetrahedral site is regular, while in these spinels the Jahn–Teller effect is expressed as a considerable flattening of the CuO₄ tetrahedron along the c axis (see Fig. 3). The width-to-height ratio of the CuO₄ coordination polyhedron is 1.26 in CuCr₂O₄ and 1.33 in CuRh₂O₄. This distortion is produced by a (less severe) flattening of the unit cell, coupled with movement of the O atoms in such a manner as to increase the CuO₄ group distortion while keeping the CrO₆ or RhO₆ octahedra quite regular. The cell distortion results in a slightly closer approach of nearest Cr (or Rh) atoms along some of the chains of octahedra. This is more destabilizing with a smaller cell and probably accounts for the Jahn–Teller distortion being slightly

less in the chromium spinel than in the larger-volume rhodium spinel.

Restudy of these spinels concludes that simple axially symmetric distortion of the spinel structure lowers the symmetry from $Fd\bar{3}m$ to $I4_1/amd$ irrespective of whether the sites are axially compressed or axially extended. The particular choice of space group is also significant in assigning the symmetry of JT spinels further transformed by lowering of temperature or by solid solution (Armbruster, Lager, Ihringer, Rotella & Jorgensen, 1983; Ivanov & Talanov, 1995).

Experimental

CuCr₂O₄ was prepared by heating a pressed pellet of a mixture of Cr₂O₃ and CuO in air for 68 h at 1173 K. A minor amount of Cu²⁺ is reduced under this oxygen fugacity producing CuCrO₂ and Cr₂O₃, measured by Rietveld refinement to constitute 6% and 2% of the sample, respectively. The remaining 92% is CuCr₂O₄. Rh₂O₃, prepared from rhodium nitrate, was mixed with CuO, pressed into a pellet and heated at 1373 K in a vertical tube furnace under flowing O₂ for ca 96 h to prepare the CuRh₂O₄ sample. Minor reduction of Cu²⁺ produced CuRhO₂, measured by Rietveld refinement to constitute 4% of the sample. No Rh₂O₃ was detected. The remaining 96% is CuRh₂O₄.

CuCr₂O₄

Crystal data

CuCr₂O₄
 $M_r = 231.55$
 Tetragonal
 $I4_1/amd$
 $a = 6.0341(4)$ Å
 $c = 7.7888(6)$ Å
 $V = 283.60(4)$ Å³
 $Z = 4$
 $D_x = 5.43$ Mg m⁻³
 D_m not measured

Cu $K\alpha$ radiation
 $\lambda = 1.54184$ Å
 $\theta = 14-73^\circ$
 $\mu = 70.7$ mm⁻¹
 $T = 295$ K
 $15 \times 12 \times 0.1$ mm
 Brown

Data collection

Philips diffractometer
 $\theta/2\theta$ scans
 Specimen mounted in reflection mode

$2\theta_{\min} = 28^\circ$
 $2\theta_{\max} = 146^\circ$
 Increment in $2\theta = 0.025^\circ$

Refinement

$R_p = 0.028$
 $R_{wp} = 0.037$
 $R_{\text{exp}} = 0.027$
 $R_B = 0.039$
 Excluded region(s): none
 Profile function: pseudo-Voigt

22 parameters
 $(\Delta/\sigma)_{\max} = 0.10$
 Preferred orientation correction: none
 Scattering factors from *International Tables for X-ray Crystallography* (Vol. IV)

Table 1. Fractional atomic coordinates and isotropic displacement parameters (Å²) for CuCr₂O₄

	x	y	z	U_{iso}
Cu	0	1/4	7/8	0.006 (2)
Cr	0	1/2	1/2	0.003 (2)
O	0	0.4672 (9)	0.2480 (7)	0.001 (2)

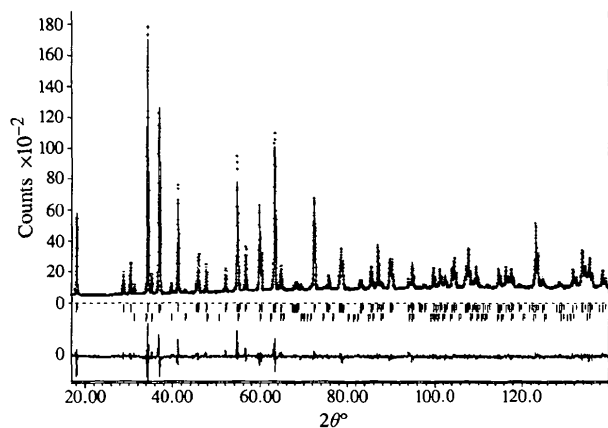


Fig. 2. Observed (crosses), calculated (solid line) and difference profiles from Rietveld analysis of the rhodium spinel sample. Tick marks indicate the positions of the allowed reflections for CuRh₂O₄ (upper), and CuRhO₂ (lower).

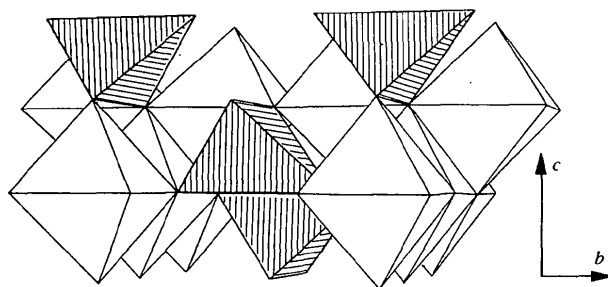


Fig. 3. CuCr₂O₄ viewed nearly along the a axis showing the compressed CuO₄ tetrahedra and the chains of edge-sharing CrO₆ octahedra.

Table 2. Selected geometric parameters (\AA , $^\circ$) for CuCr_2O_4

Cu—O ⁱ	1.957 (5) \times 4	Cr—O ⁱⁱ	1.999 (4) \times 4
Cr—O	1.973 (5) \times 2		
O ⁱ —Cu—O ⁱⁱⁱ	121.4 (1)	O—Cr—O ⁱⁱ	86.1 (3)
O ⁱ —Cu—O ^{iv}	103.9 (5)	O—Cr—O ^v	93.9 (3)
O ^v —Cr—O ^{vi}	82.0 (2)	O ⁱⁱ —Cr—O ^{vi}	98.0 (2)

Symmetry codes: (i) $-x, 1 - y, 1 - z$; (ii) $y - \frac{1}{4}, \frac{1}{4} - x, \frac{3}{4} - z$; (iii) $x, y - \frac{1}{2}, 1 - z$; (iv) $y - \frac{3}{4}, \frac{1}{4} + x, \frac{3}{4} + z$; (v) $\frac{1}{4} - y, \frac{3}{4} + x, \frac{1}{4} + z$; (vi) $y - \frac{1}{4}, \frac{3}{4} + x, \frac{1}{4} + z$.

Table 3. Contact distances (\AA) for CuCr_2O_4

Cr...Cr ⁱ	2.8884 (2)	Cr...Cr ⁱⁱ	3.0170 (2)
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Symmetry codes: (i) $y - \frac{1}{4}, \frac{1}{4} - x, \frac{3}{4} - z$; (ii) $x, y - \frac{1}{2}, 1 - z$.

CuRh₂O₄*Crystal data***CuRh₂O₄** $M_r = 333.36$

Tetragonal

 $I4_1/amd$ $a = 6.1756 (5) \text{\AA}$ $c = 7.9006 (6) \text{\AA}$ $V = 301.31 (4) \text{\AA}^3$ $Z = 4$ $D_x = 7.35 \text{ Mg m}^{-3}$ D_m not measured*Data collection*

Philips diffractometer

 $\theta/2\theta$ scans

Specimen mounted in reflection mode

Cu K α radiation $\lambda = 1.54184 \text{\AA}$ $\theta = 9-70^\circ$ $\mu = 96.8 \text{ mm}^{-1}$ $T = 295 \text{ K}$ $15 \times 12 \times 0.1 \text{ mm}$

Black

 $2\theta_{\min} = 17^\circ$ $2\theta_{\max} = 140^\circ$ Increment in $2\theta = 0.025^\circ$ *Refinement* $R_p = 0.044$ $R_{wp} = 0.060$ $R_{\text{exp}} = 0.028$ $R_B = 0.047$

Excluded region(s): none

Profile function: pseudo-Voigt

20 parameters

 $(\Delta/\sigma)_{\max} = 0.10$

Preferred orientation correction: none

Scattering factors from *International Tables for X-ray Crystallography* (Vol. IV)Table 4. Fractional atomic coordinates and isotropic displacement parameters (\AA^2) for CuRh_2O_4

	x	y	z	U_{iso}
Cu	0	1/4	7/8	0.016 (2)
Rh	0	1/2	1/2	0.010 (2)
O	0	0.4693 (15)	0.2413 (13)	0.015 (2)

Table 5. Selected geometric parameters (\AA , $^\circ$) for CuRh_2O_4

Cu—O ⁱ	1.962 (8) \times 4	Rh—O ⁱⁱ	2.055 (7) \times 4
Rh—O	2.053 (8) \times 2		
O ⁱ —Cu—O ⁱⁱⁱ	124.2 (2)	O—Rh—O ⁱⁱ	87.9 (5)
O ⁱ —Cu—O ^{iv}	102.7 (8)	O—Rh—O ^v	92.1 (3)
O ^v —Rh—O ^{vi}	82.5 (3)	O ⁱⁱ —Rh—O ^{vi}	97.5 (3)

Symmetry codes: (i) $-x, 1 - y, 1 - z$; (ii) $y - \frac{1}{4}, \frac{1}{4} - x, \frac{3}{4} - z$; (iii) $x, y - \frac{1}{2}, 1 - z$; (iv) $y - \frac{3}{4}, \frac{1}{4} + x, \frac{3}{4} + z$; (v) $\frac{1}{4} - y, \frac{3}{4} + x, \frac{1}{4} + z$; (vi) $y - \frac{1}{4}, \frac{3}{4} + x, \frac{1}{4} + z$.

Table 6. Contact distances (\AA) for CuRh_2O_4

Rh...Rh ⁱ	2.9442 (3)	Rh...Rh ⁱⁱ	3.0878 (3)
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Symmetry codes: (i) $y - \frac{1}{4}, \frac{1}{4} - x, \frac{3}{4} - z$; (ii) $x, y - \frac{1}{2}, 1 - z$.

For both compounds, program(s) used to refine structures: *DBWS-9411* (Young, Sakthivel, Moss & Paiva-Santos, 1994); molecular graphics: *ATOMS* (Dowty, 1989).

Lists of powder diffraction intensity data have been deposited with the IUCr (Reference: BR1169). Copies may be obtained through The Managing Editor, International Union of Crystallography, 5 Abbey Square, Chester CH1 2HU, England.

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Acta Cryst. (1997). **C53**, 659–661**Pentaamminechlorocobalt(III)****Hydroxotetranitronitrosylruthenate(III)**VITALY K. BELSKY^a AND VERA N. KOKUNOVA^b

^aL. Karpov Institute of Physical Chemistry, 10 Vorontsovo pole, 103064 Moscow, Russia, and ^bMoscow Academy of Fine Chemical Technology, 86 Prospekt Vernadskogo, 117571 Moscow, Russia. E-mail: belsky@cc.nifhi.ac.ru

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Abstract

The structure of $[\text{CoCl}(\text{NH}_3)_5][\text{Ru}(\text{NO}_2)_4(\text{OH})(\text{NO})]$ consists of octahedral $[\text{CoCl}(\text{NH}_3)_5]^{2+}$ and $[\text{Ru}(\text{NO}_2)_4(\text{OH})(\text{NO})]^{2-}$ ions. These units are held together via a three-dimensional network of intermolecular $\text{O—H}\cdots\text{O}$, $\text{N—H}\cdots\text{O}$ and $\text{N—H}\cdots\text{Cl}$ hydrogen bonds.